

The Infrared  $\nu_5$  Band (HNC Bend) of Protonated Hydrogen Cyanide,  $\text{HCNH}^+$ 

The first spectroscopic observation of  $\text{HCNH}^+$  was reported by Altman, Crofton, and Oka three years ago (1, 2). They used a difference frequency laser system to observe the  $\nu_1$  (NH stretch) and the  $\nu_2$  (CH stretch) bands in an A.C. glow discharge of  $\text{H}_2$  and HCN. On the basis of this result, millimeter-wave rotational transitions of  $\text{HCNH}^+$  were observed by Bogey, Demuynck, and Destombes (3) in the laboratory and by Ziurys and Turner (4) in interstellar space. More recently, Amano and Tanaka utilized the high ion concentration in a hollow cathode discharge cell to observe isotopic species and the hot bands  $\nu_4 + \nu_1 \leftarrow \nu_4$  and  $\nu_3 + \nu_1 \leftarrow \nu_3$  and thus determined rotational constants in these states (5). Tanaka, Kawaguchi, and Hirota used a diode laser to study the  $\nu_4$  fundamental (HCN bend) in the 12- $\mu\text{m}$  region (6). In this paper we report our observation of the  $\nu_5$  fundamental band (HNC bend) using a diode-laser system.

The apparatus is similar to that used for the study of the  $\nu_2$  band of  $\text{H}_3\text{O}^+$  (7).  $\text{HCNH}^+$  was produced in

TABLE I

Observed Wavenumbers of the  $\nu_5$  Fundamental Band of  $\text{HCNH}^+$  ( $\text{cm}^{-1}$ )

Transition	Wavenumber <sup>a</sup>
P(23)	587.259 (2)
P(22)	589.774 (-2)
P(16)	604.850 (-1)
P(13)	612.361 (-2)
P(12)	614.860 (-3)
P(8)	624.846 (6)
P(7)	627.328 (-1)
Q(2)	644.711 (0)
Q(3)	644.735 (-1)
Q(4)	644.769 (-1)
Q(5)	644.811 (-1)
Q(6)	644.862 (-1)
Q(7)	644.924 (3)
Q(8)	644.988 (-1)
Q(9)	645.064 (0)
Q(11)	645.243 (2)
Q(12)	645.343 (1)
Q(13)	645.450 (-1)
Q(26)	647.627 (-2)
Q(27)	647.854 (0)
Q(28)	648.089 (1)
R(7)	664.364 (-3)
R(10)	671.702 (1)
R(11)	674.140 (0)
R(12)	676.578 (3)
R(14)	681.439 (1)

<sup>a</sup> Values in parentheses are (observed - calculated)  $\times 10^3$ .

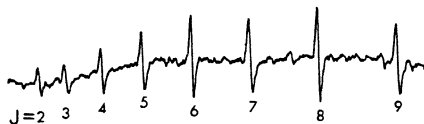


FIG. 1. The first eight  $Q$ -branch lines of the  $\nu_5$  fundamental band of  $\text{HCNH}^+$ . A time constant of 3 sec was used.

a water-cooled A.C. discharge cell (1.2 cm i.d., 1 m in length) using a mixture of HCN and  $\text{H}_2$  ( $\sim 1:5$ ) with a total pressure of 1.5 to 2 Torr, a voltage of  $\sim 4$  kV, and a current of  $\sim 100$  mA. The velocity modulation technique of Gudeman, Saykally, and others (8) was used to discriminate the ion signals from the stronger neutral signals and to increase the sensitivity of detection. The signal-to-noise ratio of absorption lines was considerably lower (by a factor of  $\sim 5$ ) in an air-cooled cell. Three mesa-stripe geometry diodes were used to cover the region from 500 to 700  $\text{cm}^{-1}$ , where the  $\nu_5$  band was theoretically predicted (9–11).

A total of 26 transitions were observed. Figure 1 shows the observed spectral lines of the first eight  $Q$  branch lines ( $Q(2)$ – $Q(9)$ ). The measured wavenumbers of the lines are listed in Table I. The spectrum was fitted using least-squares method to the energy expression for the excited state:

$$E' = B'[J(J+1) - 1] - D'[J(J+1) - 1]^2 \pm \frac{1}{2}[q + q_J J(J+1)]J(J+1),$$

where  $q$  and  $q_J J(J+1)$  are the  $\ell$ -doubling constant and its rotational dependence, respectively. The ground state rotational constants  $B''$  and  $D''$  were fixed at the values given by Bogey *et al.* (3). The determined molecular constants are listed in Table II together with the values obtained by Amano and Tanaka (5). The theoretical predictions by Lee and Schaefer (9), DeFrees and McLean (10), and Botschwina (11) are given in the footnote of the table. The observed value of the  $\ell$ -doubling constant is very close to the approximated value (12)  $q \sim 2.6B''^2/\omega_5$ .

TABLE II

Molecular Constants of  $\text{HCNH}^+ \nu_5$  (in  $\text{cm}^{-1}$ )<sup>a</sup>

	This Result <sup>b</sup>	Ref 5
$\nu_0$	645.9233(88) <sup>c</sup>	
$B'$	1.237 507(10)	1.237 527(15)
$D'$	$1.619(15) \times 10^{-6}$	$1.644(67) \times 10^{-6}$
$q$	0.005 492(11)	0.005 480(29)
$q_j$	$-2.7(20) \times 10^{-8}$	$1.1(13) \times 10^{-7}$

<sup>a</sup> Quoted errors are  $1\sigma$ .

<sup>b</sup> Fitting is done with  $B''$  and  $D''$  fixed at 1.236046 and  $1.6068 \times 10^{-6} \text{ cm}^{-1}$ , respectively.

<sup>c</sup> Theoretical prediction of  $\nu_0$  (HNC bend) ( $\text{cm}^{-1}$ ): Lee *et al.* (9), 700; DeFrees and McLean (10), 699; Botschwina (11), 620.

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## REFERENCES

1. R. S. ALTMAN, M. W. CROFTON, AND T. OKA, *J. Chem. Phys.* **80**, 3911 (1984).
2. R. S. ALTMAN, M. W. CROFTON, AND T. OKA, *J. Chem. Phys.* **81**, 4255 (1984).
3. M. BOGEY, C. DEMUYNCK, AND J. L. DESTOMBES, *J. Chem. Phys.* **83**, 3703 (1985).
4. L. M. ZIURYS AND B. E. TURNER, *Astrophys. J.* **302**, L31 (1986).
5. T. AMANO AND K. TANAKA, *J. Mol. Spectrosc.* **116**, 112 (1986).
6. K. TANAKA, K. KAWAGUCHI, AND E. HIROTA, *J. Mol. Spectrosc.* **117**, 408 (1986).
7. D. J. LIU, N. N. HAESE, AND T. OKA, *J. Chem. Phys.* **82**, 5368 (1985).
8. C. S. GUDEMAN, M. H. BEGEMANN, J. PFAFF, AND R. J. SAYKALLY, *Phys. Rev. Lett.* **50**, 727 (1983).
9. T. J. LEE AND H. F. SCHAEFER III, *J. Chem. Phys.* **80**, 2977 (1984).
10. D. J. DEFREES AND A. D. MCLEAN, *J. Chem. Phys.* **82**, 333 (1985).
11. P. BOTSCHWINA, *Chem. Phys. Lett.* **124**, 382 (1986).
12. C. H. TOWNES AND A. L. SCHAWLOW, "Microwave Spectroscopy" McGraw-Hill (1955).

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